

QUESTIONS & ANSWERS

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Answer: B

QUESTION: 600

Which of the following industries did not experience major growth from the late 1800s to the early 1900s?

- A. Beef
- B. Steel
- C. Oil
- D. Cloth
- E. Wood

Answer: A

QUESTION: 601

Which statement does not accurately portray changes to the American labor force wrought by industrialization in the late 19th and early 20th centuries?

- A. By 1880, there were almost five million individuals working in industrial jobs.
- B. More than two million women were employed in factories or offices by 1880.
- C. By the year 1890, 25% of children aged 10 to 15 years were in the workforce.
- D. During these years it was common for workers to sustain injuries and illness.
- E. Workers' compensation, disability, and retirement did not exist for workers.

Answer: C

QUESTION: 602

Which of the following is true regarding business consolidation around the turn of the 19th and 20th centuries in American industries?

- A. John D Rockefeller used vertical integration with his Standard Oil Company.
- B. Gustavus Swift was the first to use horizontal integration with meat packing.
- C. Social Darwinists were opposed to consolidation as well as to deregulation.
- D. Economists against consolidation favored more control of natural processes.
- E. The Sherman Anti-Trust Act proved quite effective in preventing monopoly.

Answer: D

QUESTION: 603

Which of the following statistics is not accurate with respect to the urbanization of America between 1870 and 1920?

- A. The number of Americans who lived in cities grew more than fivefold during this time.
- B. By 1920, almost half of the American population lived in cities with over 2500 people.
- C. Cities having populations greater than 100 000 went up by more than 3.5 times.
- D. Cities having populations greater than 500 000 increased to six-fold.

Answer: B

QUESTION: 604

Around the turn of the 19th and 20th century, when city slums resulted from overcrowding due to immigration, in which of the following areas was the least improvement made?

- A. Housing
- B. Sewage
- C. Firefighting
- D. Street lights
- E. Water supply

Answer: A

QUESTION: 605

Which of the following was not a development of city culture in America around the turn of the 19th and 20th century?

- A. Baseball had become America's favorite professional sport by the 1880s.
- B. Vaudeville shows rode trains from cities to small towns across America.
- C. Early directors like W Griffith developed the motion picture industry.
- D. Pulitzer and Hearst gave mass popularity to newspapers and magazines.
- E. All of these were American urban cultural developments during this time.

Answer: E

Section 48: Sec Forty Eight (606 to 614)
Details: High Pressure Chemistry Practice

QUESTION: 606

What is the pressure of a mixture of CO_2 , SO_2 and H_2O gases, if each gas has a partial pressure of 250 torr?

- A. 250 torr
- B. 500 torr
- C. 750 torr
- D. 1,000 torr

Answer: C

QUESTION: 607

A gas sample has a volume of 25.0 milliliters at a pressure of 1.0 atmosphere. If the volume increases to 50.0 milliliters and the temperature remains constant, the new pressure will be

- A. 1.0 atm
- B. 2.0 atm
- C. .250 atm
- D. .500 atm

Answer: D

QUESTION: 608

An increase of the temperature of a system at equilibrium favors the

- A. Endothermic reaction and decreases its rate
- B. Endothermic reaction and increases its rate
- C. Exothermic reaction and decreases its rate
- D. Exothermic reaction and increases its rate

Answer: B

QUESTION: 609

A sample of oxygen gas in a closed system has a volume of 200 milliliters at 600 K. If the pressure is held constant and the temperature is lowered to 300 K, the volume of the gas will be

- A. 100 ml
- B. 200 ml
- C. 300 ml
- D. 400 ml

Answer: A

QUESTION: 610

An assumption of the kinetic energy of gases is that the particles of a gas have

- A. Little attraction for each other and a significant volume
- B. Little attraction for each other and an insignificant volume
- C. Strong attraction for each other and a significant volume
- D. Strong attraction for each other and an insignificant volume

Answer: B

QUESTION: 611

What is the volume occupied by 2.0 moles of Ar(g) at STP?

- A. 22.4 L
- B. 44.8 L
- C. 89.6 L
- D. 179 L

Answer: B

QUESTION: 612

Which gas is least likely to obey the ideal gas laws at very high temperatures and very low temperatures?

- A. Kr

- B. Ne
- C. He
- D. Xe

Answer: D

QUESTION: 613

A gas at STP has a volume of 1.0 liters. If the pressure is doubled and the temperature remains constant, the new volume of the gas will be

- A. .25 L
- B. 2.0 L
- C. .50 L
- D. 4.0 L

Answer: C

QUESTION: 614

At 1 atmosphere of pressure, 25.0 grams of a compound at its boiling point is converted to a gas by the addition of 8,180 calories. What is the heat of vaporization for this compound, in calories per gram?

- A. 25.0 cal/g
- B. 327 cal/g
- C. 540 cal/g
- D. 8,140 cal/g

Answer: B

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